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Water molecules (H2O) consist of three atoms: A slightly negatively charged oxygen atom (O) binds two slightly positively charged hydrogen atoms (H) each. In the liquid or gaseous state (as water vapor), the molecules move freely in space. When the temperature drops and water freezes into ice, the individual water molecules bond with one another. And indeed very symmetrically, in the form of hexagons, which grow through the attachment of further molecules into a three-dimensional hexagonal lattice. Chemists call this resulting structure a "hexagonal crystal system". Prof. Dr. Udo Buck from the Max Planck Institute for Dynamics and Self-Organization says: "We recently discovered in an experiment that just 475 water molecules already form an ice crystal; before that, around 1000 molecules were assumed to be the minimum for a complete crystal." Through the attachment of ever more water molecules, more and more ice crystals grow under supercooling. From: https://t.me/truuinfochannel

It's good.
to be truu.